

Rerefinement of decabromodiphenyl ether from new area-detector data

Anthony A. Mrse *et al.*

Electronic paper

This paper is published electronically. It meets the data-validation criteria for publication in Acta Crystallographica Section C. The submission has been checked by a Section C Co-editor though the text in the 'Comments' section is the responsibility of the authors.

Rerefinement of decabromodiphenyl ether from new area-detector data

Anthony A. Mrse, Steven F. Watkins and Frank R. Fronczek*

Department of Chemistry, Louisiana State University, Baton Rouge, LA 70803-1804, USA

Correspondence e-mail: fronz@chxray1.chem.lsu.edu

Received 26 September 2000

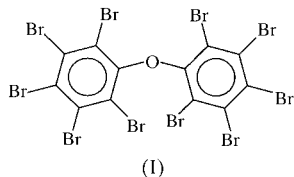
Accepted 30 October 2000

Data validation number: IUC0000318

Refinement of the title compound, bis(pentabromophenyl) ether, $C_{12}Br_{10}O$, from 100 K CCD data yields physically reasonable anisotropic displacement parameters for the C atoms. The C—Br distances are in the range 1.867 (4)–1.891 (4) Å and the dihedral angle formed by the phenyl rings is 78.3 (2)°. All endocyclic C—C—C angles are within 0.8 (4)° of the ideal value of 120°.

Comment

Ericksson *et al.* (1999) recently reported the structure of the title flame-retardant compound, (I), at 100 K. They were unable to refine the C atoms with anisotropic displacement parameters, as some became non-positive definite. They observed an asymmetry in the O—C—C angles to both rings, as well as a spread of 117.3 (5)–122.0 (4)° among the endocyclic angles within phenyl ring C1–C6, but no such asymmetry in phenyl ring C7–C12. The asymmetry was attributed to intermolecular Br···Br contacts.



With serial diffractometer (CAD-4) data at both 100 and 150 K, we likewise experienced problems treating the C atoms of the title compound as anisotropic and observed that angles O1—C6—C5 and O1—C7—C12 were 5–10° larger than O1—C6—C1 and O1—C7—C8, but did not observe the asymmetric endocyclic bond angles. We have now refined the structure *versus* CCD data at 100 K, which yielded physically reasonable ellipsoids for the C atoms. Further, all endocyclic C—C—C angles are within 0.8° (2σ) of the ideal value of 120°. We do

find the asymmetry in the O—C—C angles, however, it is also present in 1-bromo-4-(4-bromophenoxy)benzene (Mrse *et al.*, 1998), and is thus unrelated to the perbromination of the title compound.

It is noteworthy that the smallest endocyclic C—C—C angle found by Ericksson *et al.* (1999) is at the C atom (C2) with the shortest C—Br distance, and likewise the largest C—C—C angle at a brominated site is at C3, which has the longest C—Br distance. This implies that the observed spread in C—C—C angles is likely a result of slightly misplaced C atoms, or of underestimation of the uncertainties in the C positions.

The cell dimensions of the title compound at 150 K are $a = 8.3194$ (8), $b = 8.9602$ (7), $c = 14.3638$ (11) Å, $\alpha = 80.368$ (7), $\beta = 84.494$ (8), $\gamma = 63.140$ (8)°; $V = 941.5$ (2) Å³; those at 297 K are $a = 8.4295$ (13), $b = 9.0108$ (6), $c = 14.4141$ (15) Å, $\alpha = 80.742$ (11), $\beta = 85.009$ (14), $\gamma = 62.848$ (10)°; $V = 961.4$ (3) Å³.

Experimental

The title compound was kindly provided by the Albemarle Corporation. Single crystals were obtained by slow evaporation of a solution in dichlorobenzene.

Crystal data

$C_{12}Br_{10}O$	$Z = 2$
$M_r = 959.22$	$D_x = 3.393$ Mg m ⁻³
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 8.3070$ (5) Å	Cell parameters from 12 388 reflections
$b = 8.9554$ (4) Å	$\theta = 2.5$ – 40.2°
$c = 14.3472$ (8) Å	$\mu = 21.34$ mm ⁻¹
$\alpha = 80.407$ (3)°	$T = 100$ K
$\beta = 84.441$ (2)°	Prism, colourless
$\gamma = 63.194$ (2)°	$0.10 \times 0.07 \times 0.05$ mm
$V = 938.97$ (9) Å ³	

Data collection

KappaCCD diffractometer (with Oxford Cryosystems Cryostream)	10 939 independent reflections
ω scans with κ offsets	4884 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (HKL SCALEPACK; Otwinowski & Minor, 1997)	$R_{int} = 0.095$
$T_{min} = 0.140$, $T_{max} = 0.344$	$\theta_{max} = 40.2^\circ$
12 388 measured reflections	$h = 0 \rightarrow 15$
	$k = -14 \rightarrow 15$
	$l = -25 \rightarrow 26$
	Intensity decay: none

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0227P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.043$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.085$	$(\Delta/\sigma)_{max} = 0.001$
$S = 0.892$	$\Delta\rho_{max} = 2.01$ e Å ⁻³
10939 reflections	$\Delta\rho_{min} = -1.81$ e Å ⁻³
209 parameters	Extinction correction: SHELXL97
	Extinction coefficient: 0.00156 (12)

The coordinates of Ericksson *et al.* (1999) were used as a starting model, and were translated by $x + \frac{1}{2}$ to avoid most negative values. The largest final difference peak is located 0.78 Å from Br7, and the deepest hole is located 0.55 Å from Br10.

Data collection: COLLECT (Nonius, 1999); cell refinement: HKL SCALEPACK (Otwinowski & Minor, 1997); data reduction: HKL DENZO and SCALEPACK (Otwinowski & Minor, 1997); program(s) used to solve structure: coordinates of Ericksson *et al.* (1999); program(s) used to refine structure: SHELXL97 (Sheldrick,

1997); software used to prepare material for publication: *SHELXL97*.

The purchase of the diffractometer was made possible by Grant No. LEQSF(1999-2000)-ESH-TR-13, administered by the Louisiana Board of Regents.

References

- Eriksson, J., Eriksson, L. & Jakobsson, E. (1999). *Acta Cryst.* **C55**, 2169–2171.
- Mrse, A. A., Lee, Y., Bryant, P. L., Fronczek, F. R., Simeral, L. S. & Butler, L. G. (1998). *Chem. Mater.* **10**, 1291–1300.
- Nonius (1999). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods Enzymol.* **276**, 307–326.
- Sheldrick, G. (1997). *SHELXL97*. University of Göttingen, Germany.